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## REDUCED ALCOHOL MOUTHWASH ANTISEPTIC AND ANTISEPTIC PREPARATIONS

#### FIELD OF THE INVENTION

This invention is directed to reduced alcohol antiseptic and mouthwash compositions in which the clarity, taste and efficacy of the antiseptic mouth rinse is comparable to higher alcohol level formulations.

### BACKGROUND OF THE INVENTION

Thymol is a well known compound which is utilized for its antimicrobial activity in a variety of preparations. In particular, thymol can be utilized in oral hygiene preparations such as mouth rinses in sufficient quantities to provide desired beneficial therapeutic effects. Listerine® is a well known antiseptic mouth rinse that has been used by millions for over one hundred years and has been proven effective in killing bacteria in the oral cavity that are responsible for plaque, gingivitis and bad breath. Thymol, methyl salicylate, menthol and eucalyptol are active ingredients in antiseptic mouth rinses such as Listerine® and achieve their efficacy although present in very minute amounts. Without being restricted to any specific theory, it is now believed that efficacy and taste of antiseptic mouth rinses such as Listerine® is due to the availability or

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dissolution of these four active ingredients. Dissolution is also important from an aesthetic point of view in that a clear amber-colored mouthwash solution is certainly preferred by consumers to one that is cloudy or turbid or heterogeneous.

United States Patent No. 4.945.087 to Talwar et al. discloses an oral antiseptic composition containing thymol. eucalyptol. menthol and methyl salicylate wherein the unpleasant. medicinal taste of thymol is masked using effective amounts of a sugar alcohol or a mixture of sugar alcohols and anethole. The ethanol level claimed is 5-35% and a level of 21.6% is disclosed.

United States Patent 4.992.259 to Schiraldi et al. discloses a mouth rinse/dentifrice composition comprising a zinc salt co-dissolved with a naturally derived anionic polymer together with thymol. methyl salicylate and menthol in 13.5% ethanol. The mouth rinse is taught to be useful in the control and prevention of bad breath and calculus.

United States Patent 4,479,673 to Iioka et al. and United States Patent No. 4,645,663 to Nakashima et al. disclose oral compositions which contain menthol and methyl salicylate or eucalyptus oil which provides cleansing efficacy for teeth that are hypersensitive. The other actives are not disclosed however.

Finally, United States Patent No. 4,130.638 to Dhabhar et al. discloses a sodium ricinoleate mouthwash composition comprising 10-25% ethanol, 0.5-2.5% of a Pluronic® surfactant

together with thymol. menthol and eucalyptol. Methyl salicylate is not present and therefore not all four of the active ingredients are present.

It is therefore an object of the present invention to provide an antiseptic mouthwash formulation that is both efficacious in terms of killing the organisms responsible for plaque, periodontal disease and bad breath while at the same time providing such efficacy using reduced levels of alcohol and maintaining the characteristic organoleptic properties of Listerine9 and other similar antiseptic-type mouth rinses. The efficacy, taste retention and clarity attributes are achieved at reduced alcohol levels of approximately 21.0% v/v by employing the surfactant levels from above 0.10% to less than 0.6% w/v. While it was believed that the non-ionic surfactant entraps a portion of the active ingredients and as such the suppressed dissolution/availability of actives would result in a reduction of efficacy as well as diminish the strong phenolic bite of the mouthwash, it was surprisingly and unexpectedly found that the addition of two co-solvents. propylene glycol and glycerin. in amounts whose combined percentage is less than the percent decrease in alcohol, enable the production of a reduced alcohol antiseptic mouthwash composition with levels of antiseptic efficacy and clarity, and desired organoleptic properties comparable to those of Listerines.

### DETAILED DESCRIPTION OF THE INVENTION

A reduced alcohol antiseptic mouthwash composition with optimal clarity and efficacy characteristics is achieved using four known active ingredients together with specific levels of surfactant and the serendipitous addition of two co-solvents. propylene glycol and glycerin.

The compositions of this invention include effective amounts of thymol and other active oils such as those selected from the group consisting of eucalyptol, menthol, methyl salicylate, and the like, and mixtures thereof. Generally, the total amount of actives present in a composition can be from about 0.05 to about 0.35% by weight, based on the weight of the composition, with about 0.16 to about 0.28% by weight of total volume of liquid oral preparation (% w/v) being preferred. For example, the compositions, as stated above, can contain eucalyptol, menthol, methyl salicylate, and thymol. Preferably the eucalyptol is present in amounts of about 0.07 to about 0.11% w/v being preferred and most preferably, from about 0.08 to about 0.10%. Menthol is preferably present in amounts of from about 0.02 to about 0.06% w/v by weight and most preferably from about 0.03 to about 0.05% w/v. Preferably, methyl salicylate is present in amounts of about 0.03 to about 0.08% by weight and most preferably from about 0.04 to about 0.07%. Finally, with respect to the actives, thymol is present in amounts of about

0.03 to about 0.09% by weight and most preferably from about 0.04 to about 0.07%, said % by weight being based on the total composition. In addition to these actives, benzoic acid is preferably present in amounts of about 0.1 to about 0.3% by weight, based on the total composition and most preferably from about 0.13 to about 0.18%.

Compositions or final products containing these active ingredients include liquid oral preparations such as a mouthwash spray or rinse. In such preparations, the vehicle, i.e. the carrier for the ingredients of the mouthwash, such as the actives, and the like, is typically a water alcohol mixture. Generally the ratio of water to alcohol is in the range of from about 3:1 to about 25:1, preferably about 3.2:1 to about 20:1 and most preferably, about 3.5:1 to about 10:1 by volume. The total amount of water-alcohol mixture in a mouthwash preparation is typically in the range from about 80% to about 99.9% by volume of the total composition.

The co-solvents which are added to effectively aid in the dissolution of the active ingredients can be present in amounts to about 3.0% v/v each, the total amount of the co-solvents not to exceed about 16% v/v. Preferably, the propylene glycol will be present in an amount of from about 1.0 to about 4.0% v/v while the glycerin will be in an amount of from about 0.2 to about 3.0% v/v or will exist in ratios of from about 20:1 to about 1:3, propylene glycol/glycerin, respectively.

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Sole additions of propylene glycol in quantities over 6.0% v/v add bitterness to the product which is offset by lesser quantities of sweeteners such as glycerin, sugar alcohols like sorbitol or artificial sweeteners like aspartame, saccharin or acesulfame. Sole additions of glycerin in amounts over 3.0% v/v add unwanted sweetness to the taste. Sole addition of propylene glycol without glycerin at 6.0% v/v or less and sole additions of glycerin at 3.0% v/v or less are not optimized and do not exhibit the same efficacious and organoleptic results but are marginally acceptable.

The pH value of such mouthwash preparations is generally from about 3.5 to about 8.0 and preferably from about 4.0 to about 7.5. A pH below 3.5 would be irritating to the oral cavity and dissolve tooth enamel. A pH greater than 8.0 would result in an unpleasant mouth feel.

Oral liquid preparations may also contain surface active agents. i.e. surfactants, in amounts up to about 5.0% and fluoride-providing compounds in amounts up to about 2.0% by weight of the preparation.

Surface active agents (surfactants) are organic materials which aid in the complete dispersion of the preparation throughout the oral cavity. The organic surface active material may be anionic, non-ionic, amphoteric, or cationic. Suitable anionic surfactants are water-soluble salts of higher fatty acid monoglyceride monosulfates, such as the sodium

salt of the monosulfated monoglycerides of hydrogenated coconut oil fatty acids: higher alkyl sulfates, such as sodium lauryl sulfate; sodium alkylether sulfates such as sodium laurylether (1-4) sulfate, alkyl aryl sulfonates, such as sodium dodecyl benzene sulfonate; higher alkyl sulfoacetates; alkyl and alkylether phosphates such as sodium laurylether (4) phosphate; higher fatty acid esters of 1,2-dihydroxy propane sulfonates; ether sulfates; alkyl phosphates; and substantially saturated higher aliphatic acyl amides of lower aliphatic amino carboxylic acids such as those having 12 to 16 carbons at the fatty acid; alkyl or acyl radicals. Examples of the last mentioned amides are N-lauroyl sarcosine, and the sodium, potassium, and ethanolamine salts of N-lauroyl, N-myristoyl or N-palmitoyl sarcosine.

The non-ionic surfactants employed are poly(oxyethylene)poly(oxypropylene) block copolymers. Such copolymers are
known commercially as poloxamers and are produced in a wide
range of structures and molecular weights with varying
contents of ethylene oxide and propylene oxide. The non-ionic
poloxamers according to the invention are non-toxic and
acceptable as direct food additives. They are stable and
readily dispersible in aqueous systems and are compatible with
a wide variety of formulating ingredients for oral
preparations. These surfactants should have an HLB
(Hydrophilic-Lipophilic Balance) of between about 10 and 30
and preferably between 10 and 25.

Thus, non-ionic surfactants useful in this invention include poloxamers:

105	188	237	334
108	215	238	335
124	217	284	338
184	234	288	407
185	235	333	

Generally these polymers should constitute from about 0.04% w/v to about 6.0% w/v by weight of total volume of liquid oral preparation (% w/v) and preferably from 0.11% to 0.17% w/v. A particularly preferred poloxamer is Poloxamer 407 having an HLB of about 22. Such a polymer is sold under the trademark Pluronic F-127® (BASF-WYANDOTTE).

Another class of non-ionic surfactants useful in this invention are ethoxylated hydrogenated castor oils. Such surfactants are prepared by hydrogenating castor oil and treating the so-formed product with from about 10 to 200 moles of ethylene glycol. They are designated as PEG (numeral) hydrogenated castor oil in accordance with the dictionary of the Cosmetics. Toiletries and Fragrance Association, 4th Ed. wherein the numeral following PEG indicates the degree of ethoxylation, i.e. the number of moles of ethylene oxide added. Suitable PEG hydrogenated castor oils include PEG 16, 20, 25, 30, 40, 50, 60, 80, 100 and 200. The ethoxylated hydrogenated castor oils are used in the same concentrations

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as the above described poly(oxyethylene) -poly(oxypropylene) block copolymers.

Other non-ionic surface active agents which may be suitable include condensates of sorbitan esters of fatty acids with from 20 to 60 moles of ethylene oxide (e.g., "Tweens"® a trademark of ICI United States, Inc.). Amphoteric agents such as quaternized imidazole derivatives and mixtures thereof may also be suitable.

Additional non-ionic surfactants which may be suitable are the condensation products of an alpha-olefin oxide containing 10 to 20 carbon atoms, a polyhydric alcohol containing 2 to 10 carbons and 2 to 6 hydroxyl groups and either ethylene oxide or a mixture of ethylene oxide and propylene oxide. The resultant surfactants are polymers having a molecular weight in the range of 400 to about 1600 and containing 40% to 80% by weight of ethylene oxide, with an alpha-olefin oxide to polyhydric alcohol mole ratio in the range of about 1:1 to 1:3.

Cationic surface active agents which may be suitable are molecules that carry a positive charge such as cetylpyridinium chloride.

preparations of this invention. These compounds may be slightly water soluble or may be fully water soluble and are characterized by their ability to release fluoride ions or fluoride containing ions in water. Typical fluoride providing

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compounds are inorganic fluoride salts such as soluble alkali metal, alkaline earth metal, and heavy metal salts, for example, sodium fluoride, potassium fluoride, ammonium fluoride, cupric fluoride, zinc fluoride, stannic fluoride, stannous fluoride, barium fluoride, sodium fluorosilicate, ammonium fluorosilicate, sodium fluorozirconate, sodium monofluorophosphate, aluminum mono- and difluorophosphate and fluorinated sodium calcium pyrophosphate.

Tin fluoride, alkaline metal fluorides and monofluorophosphates such as sodium and stannous fluoride.

sodium monofluorophosphate and mixtures thereof are preferred.

In an oral liquid preparation such as a mouthwash, the fluoride providing compound is generally present in an amount sufficient to release up to about 0.15%, preferably about 0.001% to about 0.1% and most preferably from about 0.001% to about 0.05% fluoride by weight of the preparation.

In general, the compositions of this invention are prepared utilizing techniques well known to those skilled in the art. Thus, the liquid compositions may be prepared by mixing the alcohol soluble ingredients with ethanol, adding a quantity of water to the mixture thus obtained, and then blending or mixing in the water soluble ingredients. For example, in preparing one liter of a typical liquid oral composition, thymol, eucalyptol, menthol, methyl salicylate, surfactant and benzoic acid are dissolved in and mixed with ethanol. The co-solvents, propylene glycol and glycerin are

added in ratios of from about 20:1 to about 1:3, respectively and to this resulting mixture a sufficient quantity of water is added to make up one liter.

Those skilled in the art will appreciate that the total amount of all ingredients (components) used in the compositions of this invention equals 100% by weight of the total composition. Also, unless stated otherwise, all percents herein are percent by weight of the total composition.

The following examples are illustrative only and should not be construed as limiting the invention in any way. Those skilled in the art will appreciate that variations are possible which are within the spirit and scope of the appended claims.

#### EXAMPLE 1

Example 1 sets forth four embodiments of the claimed reduced alcohol antiseptic mouthwash of the present invention.

#### TABLE I

FORMULATION A	Tamai Ouanaire Ban
Ingredient	Ingred. Quantity Per Liter of Final Solutions
1) Alcohol USP (95% Ethanol) 2) Eucalyptol 3) Thymol NF 4) Menthol USP 5) Acid Benzoic USP 6) Hydrochloric Acid (10% V/V)	227.3700 mls .9220 gms .6390 gms .4250 gms 1.5000 gms
7) Sodium Hydroxide. 10% W/V Solution 8) Methyl Salicylate NF 9) Caramel. Acid Proof 10) Poloxamer 407 11) Glycerin USP Special 12) Propylene Glycol USP 13) Water Potable	6000 gms .2150 gms 1.5000 gms 20.0000 mls 20.0000 mls QS to 1.0000 L

#### FORMULATION B

FORMULATION B	Ingred. Quantity Per
Ingredient	Liter of Final Solution
1) Alcohol USP (95% Ethanol) 2) Eucalyptol 3) Thymol NF 4) Menthol USP 5) Acid Benzoic USP 6) Hydrochloric Acid (10% V/V)	227.3700 mls .9220 qms .5390 qms .4250 qms 1.5000 qms
7) Sodium Hydroxide, 10% W/V Solution	
8) Methyl Salicylate NF 9) Caramel, Acid Proof 10) Poloxamer 407 11) Glycerin USP Special 12) Propylene Glycol USP 13) Water Potable	.6000 qms .2150 gms 1.5000 gms 5.0000 mls 15.0000 mls QS to 1.3000 L

FORMULATION C	Ingred. Quantity Per
Ingredient	Liter of Final Solutions
<ol> <li>Alcohol USP (95% Ethanol)</li> <li>Eucalyptol</li> <li>Thymol NF</li> <li>Menthol USP</li> <li>Acid Benzoic USP</li> <li>Hydrochloric Acid         <ul> <li>(10% V/V)</li> </ul> </li> <li>Sodium Hydroxide.</li> </ol>	227.3700 mls .9220 gms .6390 gms .4250 gms 1.5000 gms
10% W/V Solution  8) Methyl Salicylate NF  9) Caramel. Acid Proof  10) Poloxamer 407  11) Glycerin USP Special  12) Propylene Glycol USP  13) Water Potable	.6000 gms .2150 gms 1.4000 gms 5.0000 mls 25.0000 mls QS to 1.0000 L
FORMULATION D Ingredient	Ingred. Quantity Per Liter of Final Solutions
1) Alcohol USP (95% Ethanol 2) Eucalyptol 3) Thymol NF 4) Menthol USP 5) Acid Benzoic USP 6) Hydrochloric Acid (10% V/V) 7) Sodium Hydroxide. 10% W/V Solution	227.3700 mls .9220 gms .6390 gms .4250 gms 1.5000 gms
8) Methyl Salicylate NF 9) Caramel. Acid Proof	.6000 gms .2150 gms 1.5000 gms

The alcohol, eucalyptol, thymol, menthol, benzoic acid, methyl salicylate, poloxamer, glycerin and propylene glycol are added together and mixed until complete dissolution is achieved.

Water is then added in a sufficient quantity so that the volume is brought to 950 mls. Hydrochloric acid or sodium

1.5000 gms

50.0000 mls

QS to 1.0000 L

10) Poloxamer 407

12) Water Potable

11) Propylene Glycol USP

hydroxide is added to adjust the pH of the solution to between about 4.1 - 4.3. Both caramel and additional water are then added to add amber color and bring entire solution to a one (1) liter volume.

All examples exhibited significantly less turbidity or cloudiness at lower temperatures than standard commercially available antiseptic mouthwash. Sensory evaluation tests involving an expert taste panel also indicated none of the subjects disclosed any significant difference in the bitter, phenolic taste between these lower alcohol formulations and the standard commercially available product. Formulation D was only marginally acceptable.

#### EXAMPLE II

### REDUCED ALCOHOL LISTERINE COSOLVENT TURBIDITY STUDIES

prepared for nephelometric analysis. The samples were prepared with a 0.07% w/v Pluronic F-127® (Poloxomer 407) level to ensure turbidity levels within the optimal detection range of the Hach Ratio Turbidimeter. The lower the turbidity number, the clearer the solution. Generally levels above 10 are not acceptable for commercial purposes. Each of the samples was prepared by pipeting 22.7 ml aliquots of a stock actives/benzoic acid/ethanol solution and 5.0 ml of a 1.4% w/v aqueous Pluronic F-127® stock solution into 100 ml graduated cylinders. Before each sample was brought to 100 ml with tap

water, varying volumes of propylene glycol and glycerin were added to the samples. The samples were then analyzed with the turbidimeter. Additionally, a sample was prepared according to the -250 formula, Formulation C set forth in Example I. (with 0.14% w/v Pluronic) and analyzed similarly. The results are listed in the following table. All percentages in the table are v/v.

FORMULATION	TURBIDITY (N.T.U.)
0% propylene glycol. 0% glycerin	>200 (off scale)
0% propylene glycol. 2% glycerin	70
1% propylene glycol, 1% glycerin	20
2% propylene glycol. 0% glycerin	8.7
2% propylene glycol. 2% glycerin	5.9
2.5% propylene glycol. 0.5% glycerin	4.5
Formulation C	<0.1 (off scale)

These results indicate that the mixture of the two cosolvents reduced the level of turbidity to a greater extent than did either of the individual solvents.

#### Example III

In vitro efficacy assessments using the plaque penetration assay model have been correlated to clinical investigations. Both Listerine® and Listerine Cool Mint® have been shown to be clinically effective and both products were

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Association acceptance seal for demonstrating that each product helps prevent and reduce supragingival plaque accumulation and gingivitis when used in a conscientiously applied program of oral hygiene and regular professional care.

In vitro plaque penetration is a stringent test of the ability of an antiseptic to penetrate a dental plaque-like organic matrix and kill oral micro-organisms, a thick biofilm of <a href="Streptococcus mutans">Streptococcus mutans</a> ATCC strain 25175. The results are listed in the following table.

FORMULATION	CRITICAL KILL TIME (minutes)
A	3.5
В	3.0
C	3.4
Listerine®	3.2
Listerine Cool Mint®	3.5

The results indicate that each formulation of the two cosolvents provide comparable oral antimicrobial efficacy to higher alcohol level compositions.

#### What We Claim Is:

1) A reduced alcohol antiseptic mouthwash composition comprising an effective amount of thymol. eucalyptol, methyl salicylate and menthol dissolved in no more than 22% v/v alcohol, a surfactant, propylene glycol, glycerin, benzoic acid, and water.

- 2) The reduced alcohol antiseptic mouthwash composition of claim 1 wherein said surfactant is selected from the group consisting of anionic, non-ionic, amphoteric and cationic surface active agents.
- 3) The reduced alcohol antiseptic mouthwash compositions of claim 2 wherein said non-ionic surfactants are selected from the group consisting of poloxamers.
- 4) The reduced alcohol antiseptic mouthwash compositions of claim 3 wherein said poloxamer constitutes from 0.01 0.2 weight percent of the total mouthwash composition.
- 5) The reduced alcohol antiseptic mouthwash composition of claim 4 wherein said glycerin and propylene glycol are present in amounts up to 6.0% volume of the total mouthwash composition.

6) The reduced alcohol antiseptic mouthwash composition of claim 5 wherein said propylene glycol is present in said antiseptic mouthwash in an amount of from about 1.0% to about 4.0% v/v and said glycerin is present in an amount of from about 0.2 to about 3.0% v/v.

- 7) The reduced alcohol antiseptic mouthwash composition of claim 5 wherein the ratio of propylene glycol to glycerin present in the compositions is from about 20/1 to about 1/3, respectively.
- 8) The reduced alcohol antiseptic mouthwash of claim 7 wherein said poloxamer is Poloxamer 407.
- 9) The reduced alcohol antiseptic mouthwash of claim 8 wherein said alcohol is ethanol.
- 10) The reduced alcohol antiseptic mouthwash composition of claim 9 wherein said thymol is present in an amount of from about .04% to about .07% w/v.
- 11) The reduced alcohol antiseptic mouthwash composition of claim 10 wherein said eucalyptol is present in an amount of from about .08% to about .10% w/v.

12) The reduced alcohol antiseptic mouthwash of claim 11 wherein said menthol is present in an amount of from about .03% to about 0.05% w/v.

- 13) The reduced alcohol antiseptic mouthwash of claim 12 wherein said methyl salicylate is present in amounts of from about .04% to about .07% w/v.
- 14) A reduced alcohol antiseptic mouthwash composition consisting of no more than 22.0% v/v alcohol that is efficacious in the killing of oral microorganisms as it retains its clarity and characteristic bitter taste.
- 15) A reduced alcohol antiseptic mouthwash composition comprising an effective amount of thymol, eucalyptol, methyl salicylate and menthol dissolved in no more than 22.0% v/v alcohol, and further comprising a surfactant, propylene glycol, benzoic acid and water.
- 16) A reduced alcohol antiseptic mouthwash composition comprising an effective amount of thymol, eucalyptol, menthol, methyl salicylate, propylene glycol, glycerin, benzoic acid, an artificial sweetener and water.
- 17) The reduced alcohol antiseptic mouthwash composition of claim 16 wherein said artificial sweetener is selected from

the group consisting of aspartame, saccharin, cyclamate, acesulfame, sucralose, alitame, sugar alcohols, glycerin and mixtures thereof.

### INTERNATIONAL SEARCH REPORT

International application No. PCT/US 94/00379

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A. CLASSI	FICATION OF SUBJECT MATTER		
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C. DOCUM	MENTS CONSIDERED TO BE RELEVANT		
Category '	Citation of document, with indication, where appropriate, of the	relevant passages	Relevant to claim No.
Y	EP,A,O 497 476 (COLGATE-PALMOLI	VE) 5	1-3,7-17
	August 1992 see claims 1,4-7,15,16; example	s a,c	
			1-3,7-17
Y	EP,A,O 338 978 (WARNER-LAMBERT October 1989	CUMPANT) 23	1 3,7 17
	see claims 1-12; table 1		
	& US,A,4 945 087 (TALWAR ET AL. cited in the application	)	
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Y	JANISTYN 'Handbuch der Kosmetik Riechstoffe, band 3, Die	a und	1-3,7-17
	Körperpflegemittel'		
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